



Designation: D 721 – 06

An American National Standard



Designation: 158/69(85)

## Standard Test Method for Oil Content of Petroleum Waxes<sup>1</sup>

This standard is issued under the fixed designation D 721; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reapproval.

*This standard has been approved for use by agencies of the Department of Defense.*

### 1. Scope\*

1.1 This test method covers the determination of oil in petroleum waxes having a congealing point of 30°C (86°F) or higher as determined in accordance with Test Method **D 938**, and containing not more than 15 % of oil.<sup>2</sup>

NOTE 1—With some types of waxes, of oil contents greater than 5%, there may be an incompatibility with MEK resulting in the formation of two liquid phases. If this occurs, the test method is not applicable to the material under test.

1.2 The values stated in inch-pound units are to be regarded as standard. The values given in parentheses are mathematical conversions to SI units that are provided for information only and are not considered standard.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

### 2. Referenced Documents

2.1 *ASTM Standards:*<sup>3</sup>

**D 938** Test Method for Congealing Point of Petroleum Waxes, Including Petrolatum

**E 1** Specification for ASTM Liquid-in-Glass Thermometers

**E 128** Test Method for Maximum Pore Diameter and Permeability of Rigid Porous Filters for Laboratory Use

2.2 *Energy Institute Standards:*

Specification for IP Standard Thermometers<sup>4</sup>

### 3. Summary of Test Method

3.1 The sample is dissolved in methyl ethyl ketone, the solution cooled to –32°C (–25°F) to precipitate the wax, and filtered. The oil content of the filtrate is determined by evaporating the methyl ethyl ketone and weighing the residue.

### 4. Significance and Use

4.1 The oil content of a wax may have significant effects on several of its properties, such as strength, hardness, flexibility, scuff resistance, coefficient of friction, coefficient of expansion, melting point, and oil straining. The importance of these effects may be dependent upon the ultimate use of the wax.

### 5. Apparatus

5.1 *Filter Stick and Assembly*, consisting of a 10-mm diameter sintered glass filter stick of 10 to 15  $\mu\text{m}$  maximum pore diameter as determined by the method in **Appendix X1**, provided with an air pressure inlet tube and delivery nozzle. It is provided with a ground-glass joint to fit a 25 by 170-mm test tube. The dimensions for a suitable filtration assembly are shown in **Fig. 1**.

NOTE 2—A metallic filter stick may be employed if desired. A filter stick made of stainless steel and having a 12.7-mm (1/2-in.) disk of 10 to 15- $\mu\text{m}$  maximum pore diameter, as determined by Test Method **E 128**, has been found to be satisfactory.<sup>5</sup> The metallic apparatus is inserted into a 25

<sup>1</sup> This test method is under the jurisdiction of ASTM Committee D02 on Petroleum Products and Lubricants and is the direct responsibility of Subcommittee D02.10.0A on Physical/Chemical Properties.

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In the IP, this test method is under the jurisdiction of the Standardization Committee. This test method was issued as a joint ASTM-IP tentative in 1964.

This test method was prepared jointly by the Technical Association of Pulp and Paper Industry and ASTM International.

This test method has been adopted for use by government agencies to replace Method 5431 of Federal Test Method Standard No. 791b.

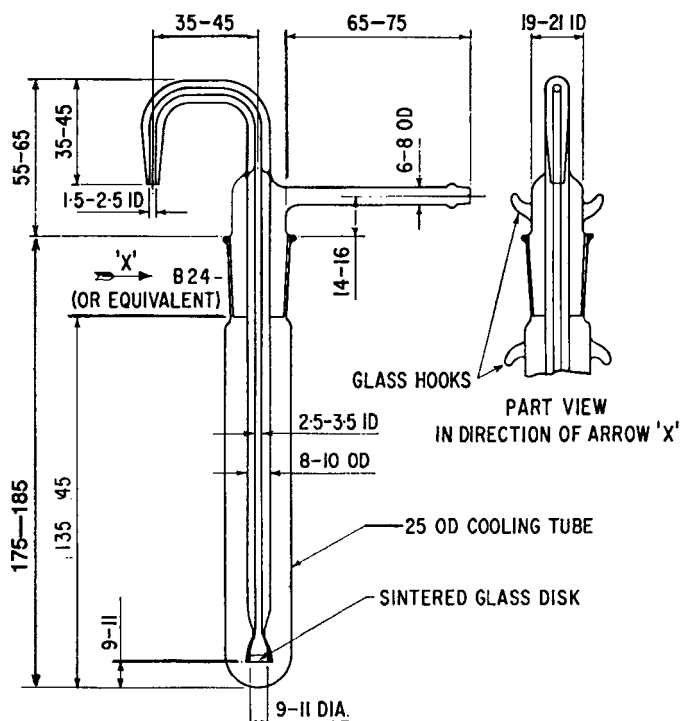
<sup>2</sup> This test method is being used by some laboratories for products of higher oil content.

<sup>3</sup> For referenced ASTM standards, visit the ASTM website, [www.astm.org](http://www.astm.org), or contact ASTM Customer Service at [service@astm.org](mailto:service@astm.org). For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

<sup>4</sup> Available from Energy Institute, 61 New Cavendish St., London, W1G 7AR, U.K.

<sup>5</sup> The sole source of supply of a suitable metal filter stick with designated porosity G known to the committee at this time is the Pall Corporation, 2200 Northern Boulevard East Hills, NY 11548. A list of United Kingdom suppliers can be obtained from Energy Institute, 61 New Cavendish St., London, W1G 7AR, United Kingdom. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,<sup>1</sup> which you may attend.

\*A Summary of Changes section appears at the end of this standard.



All dimensions are in millimetres

**FIG. 1 Filter Stick**

Temperature Range	ASTM	Thermometer Number
-37 to +21°C	71C	IP
-35 to +70°F	71F	72C
		72F

5.6.2 Temperature measuring devices other than those described in 5.6.1 are satisfactory for this test method, provided that they exhibit the same temperature response as the equivalent mercury-in-glass thermometers.

5.7 *Weighing Bottles*, glass-stoppered, having a typical capacity of 15 to 25 mL.

5.8 *Evaporation Assembly*, consisting of an evaporating cabinet and connections, essentially as illustrated in Fig. 4, and capable of maintaining a temperature of  $35 \pm 1^\circ\text{C}$  ( $95 \pm 2^\circ\text{F}$ ) around the evaporation flask. Construct the jets with an inside diameter of  $4 \pm 0.2$  mm for delivering a stream of clean, dry air vertically downward into the weighing bottle. Support each jet so that the tip is  $15 \pm 5$  mm above the surface of the liquid at the start of the evaporation. Supply purified air at the rate of 2 to 3 L/min per jet. One way to purify the air is by passage through a tube of approximately 10-mm bore packed loosely to a height of approximately 200 mm with absorbent cotton. Periodically check the cleanliness of the air by evaporating 4 mL of methyl ethyl ketone by the procedure specified in 8.5. When the residue does not exceed 0.1 mg, the evaporation equipment is operating satisfactorily.

5.9 *Analytical Balance*, capable of reproducing weights to 0.1 mg.

5.10 *Wire Stirrer*—A piece of stiff wire, made of iron, stainless steel, or Nichrome wire of about No. 20 B & S (0.9 mm in diameter) or 16 swg gage, 250 mm long. A 10-mm diameter loop is formed at each end, and the loop at the bottom end is bent so that the plane of the loop is perpendicular to the wire.

## 6. Reagents

6.1 *Methyl Ethyl Ketone*, conforming to the specifications of the Committee on Analytical Reagents of the American Chemical Society.<sup>6</sup>

6.2 Store the solvent over anhydrous calcium sulfate (5 weight % of the solvent). Filter prior to use.

6.3 *Air Supply*, clean and filtered.

NOTE 3—As an alternative, it is permissible to replace all references to “air” used in the solvent evaporation process with “nitrogen,” provided the nitrogen is clean and filtered. It should be noted, however, that the precision statements for the test method were determined using air only and that the precision associated with using nitrogen has not been determined.

## 7. Sample

7.1 If the sample of wax is 1 kg (2 lb) or less, obtain a representative portion by melting the entire sample and stirring thoroughly. For samples over 1 kg (2 lb), exercise special care

by 150-mm test tube and held in place by means of a cork.

5.2 *Cooling Bath*, consisting of an insulated box with  $30 \pm 5$ -mm ( $1.2 \pm 0.2$ -in.) holes in the center to accommodate any desired number of test tubes. The bath may be filled with a suitable medium such as kerosine, and may be cooled by circulating a refrigerant through coils, or by using solid carbon dioxide. A suitable cooling bath to accommodate three test tubes is shown in Fig. 2.

5.3 *Pipet*, or equivalent dispensing device capable of delivering  $1 \pm 0.05$  g of molten wax.

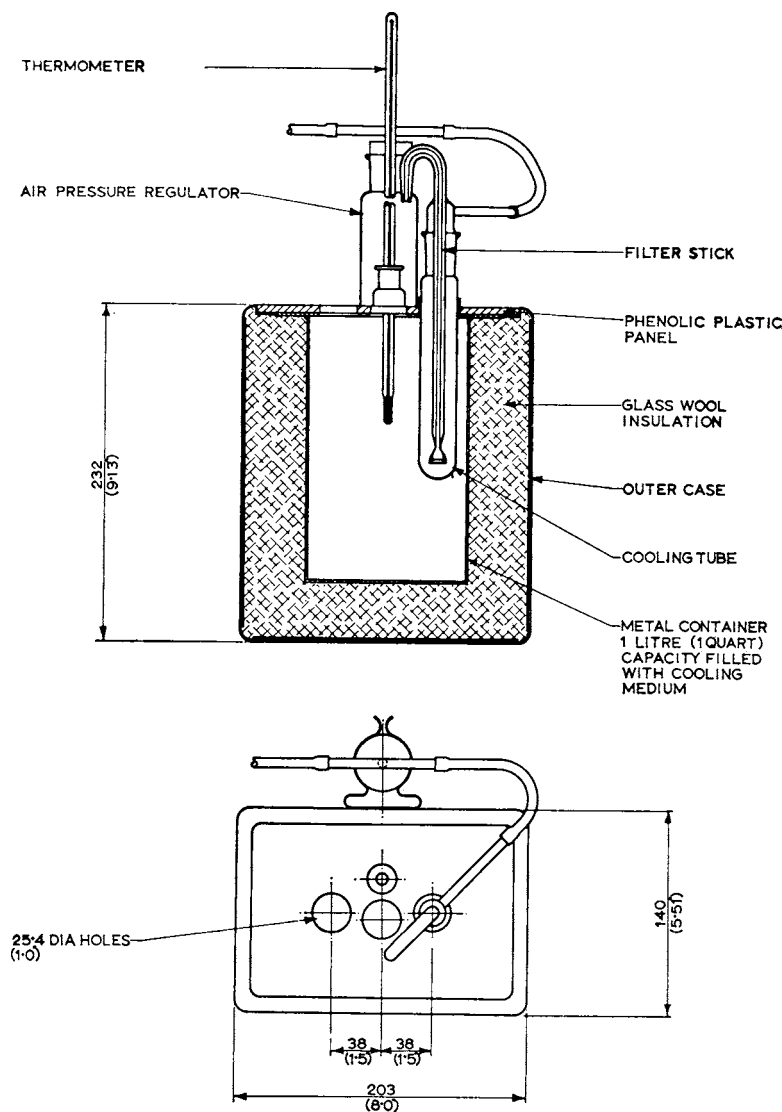
5.4 *Transfer Pipet*, or equivalent volume dispensing device, capable of delivering  $15 \pm 0.06$  mL.

5.5 *Air Pressure Regulator*, designed to supply air to the filtration assembly (8.5) at the volume and pressure required to give an even flow of filtrate. Either the conventional pressure-reducing valve or a mercury bubbler-type regulator has been found satisfactory. The latter type, illustrated in Fig. 3, consists of a 250-mL glass cylinder and a T-tube held in the cylinder by means of a rubber stopper grooved at the sides to permit the escape of excess air. The volume and pressure of the air supplied to the filtration assembly is regulated by the depth to which the T-tube is immersed in mercury at the bottom of the cylinder. Absorbent cotton placed in the space above the mercury prevents the loss of mercury by spattering. The air pressure regulatory is connected to the filter stick and assembly by means of rubber tubing.

### 5.6 Temperature Measuring Device:

5.6.1 *Thermometer*, having a range as shown below and conforming to the requirements as prescribed in Specification E 1, or in the Specification for IP Standard Thermometers.

<sup>6</sup> *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For Suggestions on the testing of reagents not listed by the American Chemical Society, see *Annual Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.



All dimensions are in millimetres (inches)

FIG. 2 Cooling Bath

to ensure obtaining a truly representative portion, bearing in mind that the oil may not be distributed uniformly throughout the sample, and that mechanical operations may express some of the oil.

**8. Procedure**

8.1 Melt a representative portion of the sample, using a water bath or oven maintained at 70 to 100°C (158 to 212°F). As soon as the wax is completely melted, thoroughly mix. Preheat the pipet or equivalent measuring device in order to prevent the solidification of wax in the tip, and withdraw a portion of the sample as soon as possible after the wax has melted. The mass of wax transferred to the test tube must be 1.00 ± 0.05 g. Allow the test tube to cool, and weigh to at least the nearest 1 mg.

NOTE 4—The weight of a test tube which is cleaned by means of solvents will not vary to a significant extent. Therefore, a tare weight may

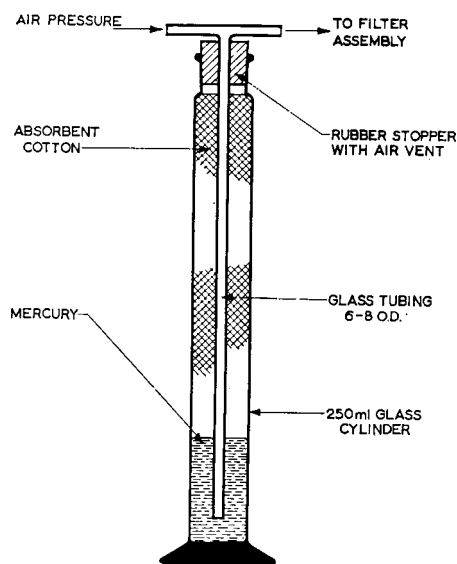
be obtained and used repeatedly.

8.2 Pipet 15 mL of methyl ethyl ketone into the test tube and place the latter just up to the level of its contents in a hot water or steam bath. Heat the solvent wax mixture, stirring up and down with the wire stirrer, until a homogeneous solution is obtained. Exercise care to avoid loss of solvent by prolonged boiling.

NOTE 5—Very high-melting wax samples may not form clear solutions. Stir until the undissolved material is well dispersed as a fine cloud.

8.2.1 Plunge the test tube into an 800-mL beaker of ice water and continue to stir until the contents are cold. Remove the stirrer. Remove the test tube from the ice bath, wipe dry on the outside with a cloth, and weigh to at least the nearest 0.1 g.

NOTE 6—During this operation the loss of solvent through vaporizations should be less than 1%. The weight of the solvent is therefore practically a constant, and after a few samples are weighed, this weight,



All dimensions are in millimetres  
**FIG. 3 Air Pressure Regulator**

approximately 11.9 g, can be used as a constant factor.

8.3 Insert the temperature measuring device into the test tube and place the test tube containing the wax-solvent slurry in the cooling bath, which is maintained at  $-34.5 \pm 1.0^\circ\text{C}$  ( $-30.0 \pm 2.0^\circ\text{F}$ ). During this chilling operation it is important that stirring by means of the temperature measuring device be almost continuous, in order to maintain a slurry of uniform consistency as the wax precipitates. Do not allow the wax to set up of the walls of cooling vessel nor permit any lumps of wax crystals to form. Continue stirring until the temperature reaches  $-31.7 \pm 0.3^\circ\text{C}$  ( $-25.0 \pm 0.5^\circ\text{F}$ ).

8.4 Remove the temperature measuring device from the tube and allow it to drain momentarily into the tube; then immediately immerse in the mixture the clean dry filter stick which has previously been cooled by placing it in a test tube and holding at  $-34.5 \pm 1.0^\circ\text{C}$  ( $-30.0 \pm 2.0^\circ\text{F}$ ) in the cooling bath for a minimum of 10 min. Seat the ground-glass joint of the filter so as to make an airtight seal. Place an unstoppered weighing bottle, previously weighed together with the glass stopper to the nearest 0.1 mg, under the delivery nozzle of the filtration assembly.

**NOTE 7**—Take every precaution to ensure the accuracy of the weight of the stoppered weighing bottle. Prior to determining this weight, rinse the clean, dry weighing bottle and stopper with methyl ethyl ketone, wipe dry on the outside with a cloth, and place in the evaporation assembly to dry for about 5 min. Then remove the weighing bottle and stopper, place near the balance, and allow to stand for 10 min prior to weighing. Stopper the bottle during this cooling period. Once the weighing bottle and stopper have been dried in the evaporation assembly, lift only with forceps. Take care to remove and replace the glass stopper with a light touch.

8.5 Apply air pressure to the filtration assembly, and immediately collect about 4 mL of filtrate in the weighing bottle. Release the air pressure to permit the liquid to drain back slowly from the delivery nozzle. Remove the weighing bottle immediately, and stopper and weigh to at least the nearest 10 mg without waiting for it to come to room temperature. Unstopper the weighing bottle and place it under one of the jets

in the evaporation assembly maintained at  $35 \pm 1^\circ\text{C}$  ( $95 \pm 2^\circ\text{F}$ ), with the air jet centered inside the neck, and the tip  $15 \pm 5$  mm above the surface of the liquid. After the solvent has evaporated, which usually takes less than 30 min, remove the bottle and stopper, and place them near the balance. Allow to stand for 10 min and weigh to the nearest 0.1 mg. Repeat the evaporation procedure, using a 5-min evaporation period instead of 30 min, until the loss between successive weighings is not over 0.2 mg.

## 9. Calculation

9.1 Calculate the amount of oil in the wax as follows:

$$\text{Oil in wax, weight \%} = (100 AC/(BD)) - 0.15 \quad (1)$$

where:

- A = weight of oil residue, g,
- B = weight of wax sample, g,
- C = weight of solvent, g, obtained by subtracting weight of test tube plus wax sample (8.1) from weight of test tube and contents (8.2),
- D = weight of solvent evaporated, g, obtained by subtracting weight of weighing bottle plus oil residue from weight of weighing bottle plus filtrate (8.5), and
- 0.15 = average factor correcting for the solubility of wax in the solvent at  $-32^\circ\text{C}$  ( $-25^\circ\text{F}$ ).

## 10. Report

10.1 Report the result as oil content, Test Method D 721. If the result is negative, report as zero.

## 11. Precision and Bias

11.1 *Precision*—The precision of this test method as determined by statistical examination of interlaboratory results is as follows:

11.1.1 *Repeatability*—The difference between two test results, obtained by the same operator with the same apparatus under constant operating conditions on identical test material, would in the long run, in the normal and correct operation of the test method, exceed the following values only in one case in twenty:

$$0.06 + 8\% \text{ of the mean}$$

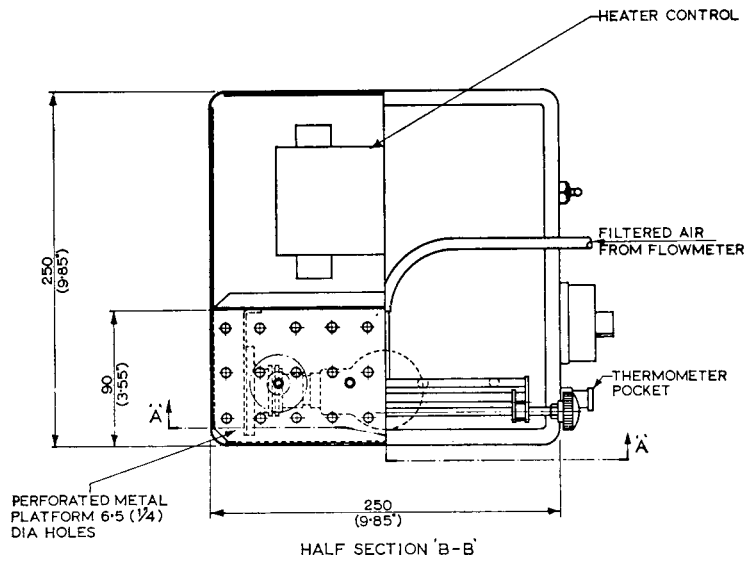
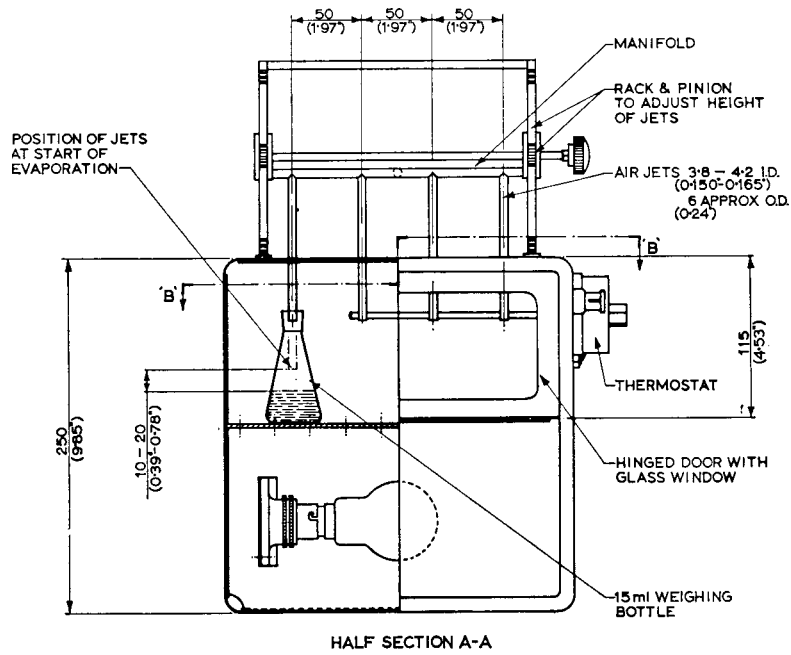
11.1.2 *Reproducibility*—The difference between two single and independent results obtained by different operators working in different laboratories on identical test material would, in the long run, in the normal and correct operation of the test method, exceed the following values only in one case in twenty:

$$0.2 + 11\% \text{ of the mean}$$

11.2 *Bias*—The procedure in this test method has no bias because the value of oil content can be defined only in terms of a test method.

## 12. Keywords

12.1 oil content; petroleum wax; wax



All dimensions are in millimetres (inches)

FIG. 4 Evaporation Assembly



APPENDIX

(Nonmandatory Information)

X1. TEST METHOD FOR MEASUREMENT OF MAXIMUM PORE DIAMETER OF RIGID POROUS FILTERS

X1.1 Scope

X1.1.1 This test method covers the determination of the acceptability of porous filter sticks used for filtration in Test Method D 721. This test method establishes the maximum pore diameter and also provides a means of detecting and measuring changes which occur from continued use.

X1.2 Definition

X1.2.1 *maximum pore diameter*—the diameter in micrometers of the largest opening in the filter.

NOTE X1.1—It is recognized that the maximum pore diameter as defined herein does not necessarily indicate the physical dimensions of the largest pore in the filter. It is further recognized that the pores are highly irregular in shape. Because of the irregularity in shape and other phenomena characteristic of filtration, a filter may be expected to retain all particles larger than the maximum pore diameter as defined and determined herein, and will generally retain particles which are much smaller than the determined diameter.

X1.3 Summary of Test Method

X1.3.1 The filter is cleaned and wetted with water. It is then immersed in water and air pressure is applied against its upper surface until the first bubble of air passes through the filter. The maximum pore diameter is calculated from the surface tension of water and the applied pressure.

X1.4 Apparatus

X1.4.1 *Pressure Measuring Device*, mercury-filled manometer or equivalent, capable of measuring pressures in increments of 0.5 mm of Hg.

X1.4.2 *Air Supply*, clean and filtered.

X1.4.3 *Air Pressure Regulator*, needle-valve type.

X1.4.4 *Drying Oven*.

X1.5 Procedure

X1.5.1 Clean the filter sticks by soaking in concentrated hydrochloric acid, and then wash them with distilled water. Rinse with acetone, air dry, and place in drying oven at 220°F (105°C) for 30 min.

X1.5.2 Thoroughly wet the clean filter to be tested by soaking it in distilled water.

X1.5.3 Assemble the apparatus as shown in Fig. X1.1. Apply pressure slowly from a source of clean air.

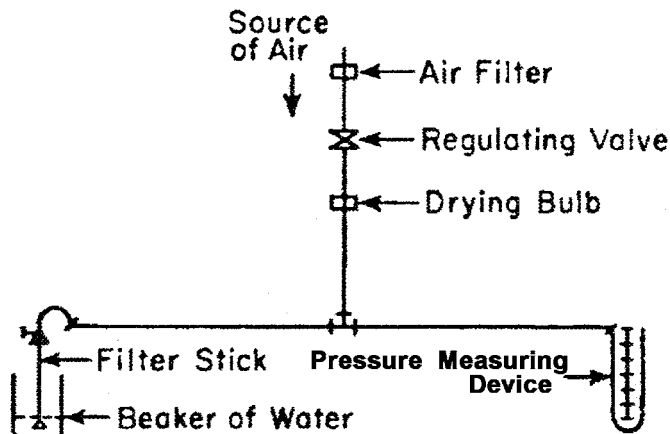


FIG. X1.1 Assembly of Apparatus for Checking Pore Diameter of Filter Sticks

X1.5.4 Immerse the filter just below the surface of the water.

NOTE X1.2—If a head of liquid exists above the surface of the filter, the back pressure produced must be deducted from the observed pressure.

X1.5.5 Increase the air pressure to about 10 mm Hg (or equivalent pressure value if using different units) below the acceptable pressure limit and then at a slow uniform rate equivalent to about 3 mm Hg/min until the first bubble passes through the filter. This can be conveniently observed by placing the beaker or test tube over a mirror. Read the pressure measuring device when the first bubble passes off the underside of the filter.

X1.6 Calculation

X1.6.1 Calculate the pore diameter as follows:

$$D = 2180/p$$

where:

$D$  = pore diameter,  $\mu\text{m}$ , and

$p$  = pressure reading, mm Hg.

NOTE X1.3—From this equation, pressure corresponding to the upper and lower limits of the specified pore diameters can be calculated. These pressures may be used for acceptance testing.

**SUMMARY OF CHANGES**

Subcommittee D02.10 has identified the location of selected changes to this standard since the last issue (D 721–05) that may impact the use of this standard.

- (1) Revised **8.1**.
- (2) Revised **8.2.1**.
- (3) Revised **8.3-8.5**.

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- (1) Corrected Eq 1.
- (2) Added **5.6.2**. Renumbered **5.6** and **5.6.1**.
- (3) Revised **5.2, 5.7, 5.9, 8.3, and 8.4**.

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